Home Search Collections Journals About Contact us My IOPscience

Ab initio investigation of phonon modes in the ${\rm MgAl}_2{\rm O}_4$ spinel

This article has been downloaded from IOPscience. Please scroll down to see the full text article. 2002 J. Phys.: Condens. Matter 14 3543 (http://iopscience.iop.org/0953-8984/14/13/312)

View the table of contents for this issue, or go to the journal homepage for more

Download details: IP Address: 171.66.16.104 The article was downloaded on 18/05/2010 at 06:24

Please note that terms and conditions apply.

Ab initio investigation of phonon modes in the MgAl₂O₄ spinel

Pascal Thibaudeau¹ and François Gervais²

 ¹ Commissariat à l'Energie Atomique Le Ripault, BP 16, F-37260 Monts, France
 ² Laboratoire d'Electrodynamique des Matériaux Avancés, FRE2077 CNRS, LRCM01 CEA, Faculté des Sciences et Techniques, Université François Rabelais, Parc de Grandmont, F-37000 Tours, France

E-mail: Pascal.Thibaudeau@cea.fr

Received 15 January 2002, in final form 22 February 2002 Published 22 March 2002 Online at stacks.iop.org/JPhysCM/14/3543

Abstract

Ab initio infrared and Raman phonon modes in the normal cubic MgAl₂O₄ spinel are calculated at the first Brillouin zone centre using density functional theory with plane-wave basis and norm-conserving pseudopotentials. Good agreement with available experimental infrared and Raman spectra is found for both natural and synthetic spinel. A new set of infrared measurements on synthetic spinels shows extra vibrational infrared modes which are not predicted by calculations on normal spinel. The ionic character of spinels is illustrated by comparison between calculated and experimental Born effective charges.

1. Introduction

Spinel oxides form an important range of ceramic compounds with great interesting electrical, mechanical, magnetic and optical properties. It is observed that wide bandgaps in these structures offer attractive photoelectronic and optical applications [1]. Studies on spinel oxides also have numerous applications in geophysics and magnetism [2, 3]. For instance, the spinel transition of olivive, which is a major constituent of Earth's mantle, is widely accepted as the origin of the near 410 km seismic discontinuity [4]. Moreover, spinel oxides are found to be highly resistant to radiation-induced swelling and transformations at elevated temperatures [5, 6]. Magnesium dialuminium oxide (MgAl₂O₄) is often considered as a prototype of the AB₂O₄ family materials, which crystallizes in spinel structure [7].

Detailed analysis of the vibrational spectra of spinels, and especially in the case of $MgAl_2O_4$, is complicated by order/disorder phenomena that occur in such structures [8]. In normal spinel, an A divalent cation occupies one of the eight possible tetrahedral sites and a B trivalent cation is located in one of the 16 possible octahedral sites. Besides, tetrahedral and octahedral cationic exchange, giving rise to partially inverse spinels [9], is experimentally observed. Such changes in the cationic order are usually induced by high temperature and/or

high pressure. According to Navrotsky and Kleppa [10], one could characterize a two-species spinel oxide by an inversion or disorder parameter x in such a way that the general formula $(A_{1-x}B_x)_{tet}[A_xB_{2-x}]_{oct}O_4$ can describe separately the tetrahedral and octahedral interstitial sites occupancies. The normal spinel is obviously found for x = 0 and the full inverse structure is given by x = 1.

From ambient to high temperatures, the disorder parameter is obtained from *in situ* neutron diffraction measurements [11]. However, due to the unquantified extent associated with the cation site exchanges [12, 13], it is difficult to describe properly the degree of short-range order and therefore to distinguish between different models of disorder. To complicate the interpretation even further, non-stoichiometric spinels AO.yB₂O₃ exist with y > 1 exhibiting cationic disorder and vacancies for both cationic sites [14, 15].

To quantify the nature and amplitude of disorder in synthetic spinels from vibrational spectra, a non-ambiguous normal mode assignment is required. Thus, an *ab initio* methodology is applied on normal MgAl₂O₄ spinel structure to clarify infrared and Raman spectroscopic interpretation. This interpretation is then discussed in the light of a new set of infrared reflectivity measurements on synthetic spinels.

2. Theoretical framework

2.1. Symmetry analysis

It is now commonly accepted that normal spinel exhibits a cubic structure, with space group $Fd3m(\Theta_h^7)$ and 56 atoms per unit cell (Z = 8). Fourfold- (special 8a Wyckoff position) and eightfold- (special 16d Wyckoff position) coordinated cationic sites are located in an oxygen (general 32e Wyckoff position) close-packed pseudo-face-centred-cubic sublattice. To complete the description of the anion position, an additional parameter, generally labelled by u and known in oxide spinels as the internal oxygen parameter, is introduced. In most spinels, u lies between 0.24 and 0.275 if the origin of the unit cell is taken at the centre of inversion. For the particular value u = 0.25, the anions form an exactly cubic close-packed lattice and define a regular tetrahedron and octahedron with the cation as a centre. As u increases, the oxygens move along the $\langle 111 \rangle$ direction, causing the tetrahedral site to enlarge at the expense of the octahedral one [16, 17].

Only two formula units are present in the primitive unit cell, leading to 42 normal modes. In addition to three acoustic modes of T_{1u} symmetry, 39 optic modes are distributed among the following symmetries at the Brillouin zone centre:

$$\Gamma = A_{1g}(R) + E_g(R) + 3T_{2g}(R) + 4T_{1u}(IR) + T_{1g} + 2A_{2u} + 2E_u + 2T_{2u}$$

where R (IR) denotes Raman (infrared) activity and the remaining modes are silent [8, 18]. In a normal, non-defective $MgAl_2O_4$ spinel and without mode couplings, selection rules suggest that only five modes should be Raman active and four modes infrared active.

2.2. Ab initio methodology

Powerful techniques were developed a few years ago to calculate vibrational properties of extended systems through density functional perturbation theory (DFPT) [19]. This theory allows self-consistent calculation of the linear response to an external perturbation in an extended system and combines the advantages of both the frozen phonon method [20–22] and the dielectric theory [23–25]. For the calculations of the electronic density response functions, it is necessary to obtain the wavefunction of the ground state only. Moreover, as there is no restriction to high-symmetry points as in the frozen phonon calculation, perturbations of

Table 1. Comparison between experimental and calculated lattice constant and oxygen parameter in the MgAl₂O₄ spinel. Relative errors are shown in parenthesis.

	Experiment ^a	DFT all-electron	DFT-LDA-NCPP (this work)
Lattice constant (Å)	8.060	8.072(+0.15%) ^b 8.056(-0.05%) ^c	7.905(-1.9%)
Internal oxygen u	0.2624	0.2748(+4.7%) ^b , 0.2638(+0.5%) ^c	0.2735(+4.2%)
^a Extrapolated to $T =$. 0.		

"Extrapolated to I = 0.

^b [37]. ^c [38].

[50].

arbitrary wavelength can be evaluated. These techniques have been successfully applied to the phonon dispersion curve of elemental and binary semiconductors [26], simple metals [27] and simple binary alkaline-earth oxides [28]. However, some phonon dispersion curves of more complex structures such as perovskites have been studied in the linear response approach within the framework of the linearized plane-wave (LAPW) method [29].

The ground state of MgAl₂O₄ spinel is calculated using density functional theory within the local density approximation (DFT-LDA) parametrized by Perdew and Wang [30]. A planewave basis set and norm-conserving pseudopotentials (NCPPs) for valence electrons only in the fully separable form given by Kleinman and Bylander [31] are used. From a scalar relativistic all-electron configuration, the pseudopotential generation follows the schemes proposed by Hamann [32] and by Troullier and Martins [33] for the cations and the oxygen respectively. Because it generates a smoother oxygen pseudopotential, the scheme derived by Troullier and Martins significantly reduces the computer memory consumption without having a noticeable effect on the results. Only linear core-valence correction is used in the generation of cationic pseudopotentials [34]. During their constructions, unphysical bound states or resonances appearing in the valence spectrum are ruled out by choosing suitable cutoff radii and local potential component as proposed by Gonze and co-workers [35]. The convergence of the pseudo-wavefunctions in momentum space is monitored to estimate a suitable basis set cutoff of 80 Ryd for the plane-wave calculations of all species. For the Brillouin zone sampling, a $4 \times 4 \times 4$ grid is used for the k-point summation in the self-consistent calculations [36], which leads to relatively unaffected structural parameters.

3. Computational results

3.1. Ground-state equilibrium properties

During self-consistent cycles, the external pressure is kept fixed. The lattice parameter and oxygen internal position are optimized according to the symmetries of the cell until the internal forces and the stress are both close to zero. The values found are reported in table 1 for zero pressure.

The pressure is then varied and the previous process is repeated until the forces acting on atoms are close to zero and the diagonal components of the stress tensor are all equal to the external pressure. Both the internal energy and volume are collected in order to fit six different equations of state (EOS) at zero temperature. All these EOSs recover the previously optimized value of lattice parameter at zero pressure up to an absolute error of 10^{-4} . To compare these calculated equilibrium properties, experimental lattice parameter and oxygen internal position are extrapolated to the zero-temperature limit from available data [39] obtained by neutron diffraction at different temperatures, neglecting the vibrational energy contribution.

	EOS	K_0 (GPa)	K_0'
DFT-LDA-NCPP	Universal Tait [40]	199	3.63
(This work)	Born-Mie [41]	198	3.56
	[42]	199	3.61
	[43]	199	3.60
	[44]	199	3.60
	3rd order		
	[45]	199	3.58
DFT-LCAO	[46]	227 ^a , 213 ^b	3.4 ^a , 5.4 ^b
Experiments		196 ^c , 196 ^d , 196 ^e	4.89 ^c , 5.3 ^d , 4.7 ^e

Table 2. Calculated bulk modulus and its first derivative with respect to the pressure compared with experimental results in the MgAl₂O₄ spinel.

^a [38].

^b [47].

^c [48]. ^d [49].

e [50].

The fitting procedure on the EOSs allows us to obtain the bulk modulus K_0 and its first derivative with respect to the pressure K'_0 as a by-product. The values calculated are summarized in table 2 and, due to the relatively small pressure extent, no obvious effect is found with a particular choice of an EOS.

However, the lowest residual χ^2 on fitted values is obtained using the Brennan–Stacey EOS. The equilibrium total energy versus density curve and the corresponding Brennan–Stacey EOS interpolation are reported in figure 1.

The lattice constant and oxygen position determined in this way are also compared with allelectron calculations. They are much more underestimated, essentially due to the linearity of the core–valence density imposed on generating cationic pseudopotentials [34]. For the same reason, the bulk modulus is slightly overestimated in the LDA compared with experiment as previously reported for other simple compounds [51] but agreement with experiments is still acceptable. Introduction of a nonlinear core–valence correction might possibly improve the results but it is out of the scope of the present paper.

3.2. Lattice dynamics at the Brillouin zone centre

The *ab initio* vibrational frequencies at the zone centre are reported in table 3 and compared with calculated values using interatomic potentials [52]. The non-analytical part of the dynamical matrix is carefully taken into account using the full Born effective charge tensors to find the correct LO–TO splitting [53]. The overall agreement between available experimental data from table 3 and interatomic potentials is very good. However little discrepancy is found for the E_g mode and one T_{2g} Raman mode around 490 cm⁻¹. The Raman intensity of the predicted T_{2g} Raman mode near 553 cm⁻¹ is probably very weak, which makes its precise experimental determination difficult.

As MgAl₂O₄ spinel is a known polar compound, the DFPT is a suitable theory which allows us to calculate the low- and high-frequency dielectric constants and the Born effective charge tensors directly on all inequivalent atoms [19]. The results are summarized in table 4. It is observed that the Born tensors are fully diagonal on cations and mainly diagonal on oxygen atoms up to 2×10^{-2} . The agreement between calculated and experimentally derived diagonal parts of the Born effective tensors is excellent.



Figure 1. Calculated EOS of normal cubic MgAl₂O₄ fitted with a Brennan–Stacey form.

Table 3. DFT-LDA-NCPP vibrational frequencies in cm^{-1} in the MgAl₂O₄ spinel compared with Lauwers and Herman results using interatomic potentials [52]. Relative errors are shown in parenthesis.

Vibrational frequencies (cm ⁻¹)							
DFT-LDA-NCPP	Lauwers and Herman						
760	772(+2%)						
380	403(+6%)						
309	281(-9%)						
553	597(+%)						
661	679(+3%)						
307	337(+10%)						
478	481(+1%)						
565	497(-12%)						
666	664(-0.3%)						
309	337(+9%)						
612	601(-2%)						
564	488(-13%)						
854	803(-6%)						
191							
495							
315							
392							
602							
669							
742							
	Vibrational fr DFT-LDA-NCPP 760 380 309 553 661 307 478 565 666 309 612 564 854 191 495 315 392 602 669 742						

The diagonal feature of the Born effective tensors is a direct consequence of the binding ionic character in the spinel. Even if each anion in that spinel is surrounded by three

 Table 4.
 Calculated low- and high-frequency dielectric constants and Born effective charges compared with various experiments.

	ϵ_0	ϵ_{∞}	Z _{Mg}	$Z_{\rm Al}$	Z _O
DFT-LDA-NCPP Experiment	8.16 8.16 ^a , 8.18–8.32 ^c	3.00 2.89–2.92 ^a , 2.93 ^b	$\begin{array}{c} 2.02\\ 2.1\pm0.4^{b} \end{array}$	$\begin{array}{c} 2.76\\ 2.7\pm0.2^{b} \end{array}$	$-1.89 - 1.89 \pm 0.04^{b}$

^a [54].

^b This work.

° [55].

octahedrally coordinated cations of Al and one tetrahedrally coordinated cation of Mg, the effective field is mainly isotropic. Based on these *ab initio* Born charge calculations, an ionic model using simple interatomic potentials should be able to accurately reproduce the main features of the vibrational spectrum at the zone centre [56]. The calculated dielectric constants are slightly overestimated by LDA to the order of several per cent but remain compatible with experiments. These overestimated values are due to the fact that LDA is widely based on the electron-gas approximation and its exchange–correlation form has a 'memory' of its metallic parentage [57].

4. Reflectivity measurements

It is a difficult task to interpret precisely the vibrational modes of $MgAl_2O_4$ in the Fd3m structure [8]. Table 5 summarizes experimental infrared and Raman vibrational frequencies for $MgAl_2O_4$ collected over the past 30 years.

Obviously, frequencies are affected by the true nature of the measured sample and many authors have tried to resolve this difficulty by comparing differently prepared samples [58]. Cationic disorder lowers the symmetry and increases available vibrational modes. As the cationic masses and total electronic charges are close in MgAl₂O₄, slight differences in disordered spinel spectra are observed. Only recently, an experimentally induced disorder Raman mode has been interpreted unambiguously [6].

Indeed, the availability of Fourier-transform infrared spectroscopy justifies remeasuring the reflectivity of the MgAl₂O₄ single crystal because resolution, spectral accuracy and sensitivity have been significantly improved with respect to earlier experiments. New measurements are carried out on two different synthetic spinel single crystals from MTI Corp, grown by the Czochralski method with two optically polished and oriented sides along the $\langle 100 \rangle$ lattice direction. The reflectivity spectra at normal incidence are collected on a Bruker IFS66v infrared Fourier spectrometer. The general results agree within experimental error with previously published spectra [54, 59, 67]. However, the very weak infrared band near 227 cm⁻¹, which is absent from previous infrared reports, is unambiguously observed in both spectra. This mode is consistent with the extra modes observed by Raman scattering as shown in table 5. It may be assigned to vibration induced by an exchange between Mg and Al sites or to the defective nature of the non-stoichiometric spinel lattice [69]. This particular point will be the subject of a forthcoming paper. The complex dielectric function at the zone centre is expressed in the form of the generalized Lyddane–Sachs–Teller relationship, first proposed by Kurosawa [61] as

$$\varepsilon(\omega) = \varepsilon_{\infty} \prod_{j=1}^{n_{IR}} \frac{\omega_j^2(\text{LO}) - \omega^2 - \iota\Gamma_j(\text{LO})\omega}{\omega_j^2(\text{TO}) - \omega^2 - \iota\Gamma_j(\text{TO})\omega},\tag{1}$$

where the ω_j and Γ_j stand for the phonon mode frequencies and dampings respectively. TO and LO suffixes refer to optically transverse and longitudinal modes respectively. Equation (1)

Ab	initio	investigation	of	phonon	modes	in the	MgAl ₂ O ₄	spinel
110	nuno	mesugation	01	phonon	moues	in the	1151 112 04	opiner

damping measurements from synthetic spinels obtained in this work.										
Reference	$A_{1g}(R)$	$E_g(R)$	$T_{2g}(R)$			$T_{1u}(IR)$				Others
[66]		410	311	492	611	305 ^a	428 ^a	485 ^a	670 ^a	722(A _{1g})
						311 ^b		497 ^b	800 ^b	
[6,58]	770	409	311		670					
[18]	767	407	312		666	304 ^a	476 ^a	578 ^a	676 ^a	800(IR)
						312 ^b	610 ^b	575 ^b	868 ^b	
[15]	769	410	310	490	668	308 ^a	494 ^a	569 ^a	676 ^a	223(R)
						313 ^b	609 ^b	570 ^b	878 ^b	725(A _{1g})
[67]						306 ^a	490 ^a	580 ^a	670 ^a	
. ,						312 ^b	609 ^b	610 ^b	857 ^b	
[68]	770	405	305	485	663					225(R)
[69]	772	410	311	492	671	305 ^a	428 ^a	485 ^a	670 ^a	727(A _{1g})
								630 ^b	855 ^b	1.5
[70]						309 ^a	522 ^a	580 ^a	688 ^a	
Frequencies (cm ⁻¹)						303 ^a	491 ^a	532 ^a	675 ^a	226 ^a (IR)
this work										805 ^a (IR)
						307 ^b	608 ^b	532 ^b	870 ^b	227 ^b (IR)
										796 ^b (IR)
Dampings (cm ⁻¹)						18 ^a	43 ^a	44 ^a	42 ^a	14 ^a
this work										74 ^a
						23 ^b	58 ^b	54 ^b	42 ^b	15 ^b
										68 ^b

Table 5. Reported experimental vibrational frequencies in cm^{-1} , including the frequencies and damping measurements from synthetic spinels obtained in this work.

^a Transverse mode.

^b Longitudinal mode.

is then combined with the Fresnel formula for reflectivity of a half-space medium near normal incidence with vacuum as reference

$$R(\omega) = \left| \frac{\sqrt{\varepsilon(\omega)} - 1}{\sqrt{\varepsilon(\omega)} + 1} \right|^2.$$
⁽²⁾

The high-frequency dielectric constant, phonon modes and dampings are then fitted from reflectivity data. An excellent agreement can be achieved as illustrated in figure 2.

Data fit parameters are compared with previous reported data in table 5. Such a combination of infrared Fourier spectrometry and data fitting allows the determination of polar phonon frequencies that are relevant in equation (1), with a precision better than 0.5%. Large dampings are found for two intense infrared modes. These large dampings associated with intense modes may be interpreted as a consequence of the grit finish as previously reported on Al₂O₃ corundum by Barker [60]. The complex index is then deduced using equation (1) combined with $\varepsilon = n^2$ and the parameters which yield the best fit to reflectivity data. The results obtained with this technique are consistent with a direct Kramers–Krönig transformation on reflectivity data.

To deduce the effective charges from measurements, a relationship proposed by Kurosawa [61] and later exploited in many compounds, mainly oxides [62–65], was used. This relation shows, still within the context of a quasi-harmonic vibrational model, that the Born effective charge Ze (where e is the electron charge) in a spherical isotropic effective field is related to TO and LO phonon frequencies via

$$\sum_{j=1}^{n_{IR}} (\omega_j^2(\text{LO}) - \omega_j^2(\text{TO})) = \frac{1}{\varepsilon_0 V} \sum_{k=1}^{n_{atom}} \frac{Z_k^2 e^2}{m_k},$$
(3)



Figure 2. Infrared reflectivity measurements and generalized Lyddane–Sachs–Teller best fitted values in the cubic MgAl₂O₄ spinel.

where the summation in the right-hand side of the equation is over all atoms of mass m_k contained in the unit cell of volume V and the summation in the left-hand side of the equation is over all IR-active modes. ε_0 is the vacuum dielectric constant. Equation (3) and the charge neutrality condition are available to determine the three Born effective charges. Therefore a realistic value for one of the cationic effective charges is required to deduce the two others. The procedure is repeated until a satisfactory effective charge set is found [15]. The oxygen effective charge is not very sensitive to the various choices of cationic charges because of the weight of the oxygen atom in equation (3). Due to the relative uncertainty of the experimental frequencies, the cationic charges may balance each other, as illustrated in table 4.

Better values of Born effective charges are found if both TO and LO infrared frequencies are correctly determined. Some authors have tried to calculate these frequencies by molecular mechanics using interatomic potentials [15, 16, 52]. Unfortunately, these potentials are often fitted on spinel experimental data, including vibrational frequencies, but frequency data themselves are often insufficient to determine accurate interatomic potentials and a full analysis on a given single crystal is not always possible.

The measured effective charge on oxygen atoms is close to the value previously found by Ishii and co-workers [15] and previously found by Gervais [63] on aluminium atoms in Al_2O_3 . However, the values calculated by Ishii on cations are different from both the present measurements and the *ab initio* calculations. These discrepancies may be due to a lack of precision in their frequency determination.

5. Conclusion

Structural properties in the normal cubic MgAl₂O₄ spinel were calculated using DFT with plane-wave basis and pseudopotentials. A DFPT vibrational mode analysis on the previously

calculated ground state was performed in order to give new insights into the interpretation of infrared and Raman spectroscopic measurements. Conclusive agreement is found for both structural and vibrational properties in the normal non-defective cubic MgAl₂O₄ spinel. A new set of infrared measurements on synthetic spinels shows modes which are forbidden by group theory for normal spinel. These infrared modes around 227 and 806 cm⁻¹ could originate in the defective nature of the spinel lattice. Calculated Raman modes on normal spinel are shown to be useful to investigate the vibrational spectrum differences on synthetic spinels, demonstrating that the Raman mode near 725 cm⁻¹ of A_{1g} character is not a harmonic natural spinel mode as experimentally observed by Cynn and co-workers [6]. Detailed analysis between calculated and measured Born effective charges suggests no significant differences between normal and synthetic spinel samples. It is shown that, even on the anion, the ionic character of binding in this spinel is recovered by the diagonal feature of the calculated Born effective tensors.

Acknowledgment

One of the authors (PT) would like to express special thanks to S Labayle-Camus for helpful discussions.

References

- Ueda N, Omata T, Hikuma N, Ueda K, Mizoquchi H, Hashimoto T and Kawazoe H 1992 Appl. Phys. Lett. 61 1954
- [2] Green H W II 1984 Geophys. Res. Lett. 11 817
- [3] Schiessl W et al 1996 Phys. Rev. B 53 9143
- [4] Ringwood A E and Major A 1970 Phys. Earth Planet. Inter. 3 89
- [5] Cain L S, Pogatshnik G J and Chen Y 1988 Phys. Rev. B 37 2645
- [6] Cynn H, Sharma S K, Cooney T F and Nicol M 1992 Phys. Rev. B 45 500
- [7] Galasso F S 1970 Structure and Properties of Inorganic Solids (New York: Pergamon)
- [8] White W B and DeAngelis B A 1967 Spectrochim. Acta A 23 985
- [9] Schmoker U, Boesch H R and Waldner F 1972 Phys. Lett. A 40 237
- [10] Navrotsky A and Kleppa O J 1967 J. Inorg. Nucl. Chem. 29 2701
- [11] Redfern S A T, Harrison R J, O'Neill H St C and Wood D R R 1999 Am. Mineral. 84 299
- [12] Wood B J, Kirkpatrick R J and Montez B 1986 Am. Mineral. 71 999
- [13] O'Neill H St C and Navrotsky A 1983 Am. Mineral. 68 181
- [14] Dupree R, Lewis M H and Smith M E 1986 Phil. Mag. A 53 L17
- [15] Ishii M, Hiraishi J and Yamanaka T 1982 Phys. Chem. Minerals 8 64
- [16] Striefler M E and Barsch G R 1972 J. Phys. Chem. Solids 33 2229
- [17] Hill R J, Craig J R and Gibbs G V 1979 Phys. Chem. Minerals 4 317
- [18] Chopelas A and Hofmeister A M 1991 Phys. Chem. Minerals 18 279
- [19] Baroni S, Giannozzi P and Testa A 1987 Phys. Rev. Lett. 58 1861
- [20] Wendel H and Martin R M 1979 Phys. Rev. B 19 5251
- [21] Kunc K and Martin R M 1982 Phys. Rev. Lett. 48 406
- [22] Yin M T and Cohen M L 1982 Phys. Rev. B 26 3259
- [23] Sham L J 1969 Phys. Rev. 188 1431
- [24] Pick P, Cohen M H and Martin R M 1970 Phys. Rev. B 1 910
- [25] Falter C 1988 Phys. Rep. 164 1
- [26] Giannozzi P, de Gironcoli S, Pavone P and Baroni S 1991 Phys. Rev. B 43 7231
- [27] de Gironcoli S 1995 Phys. Rev. B 51 6773
- [28] Schütt O, Pavone P, Windl W, Karch K and Strauch D 1994 Phys. Rev. B 50 3746
- [29] Yu R and Krakauer H 1995 Phys. Rev. Lett. 74 4067
- [30] Perdew J P and Wang Y 1992 Phys. Rev. B 45 13 244
- [31] Kleinman L and Bylander D M 1982 Phys. Rev. Lett. 48 1425
- [32] Hamann D R 1989 Phys. Rev. B 40 2980
- [33] Troullier N and Martins J L 1991 Phys. Rev. B 43 1993

- [34] Louie S G, Froyen S and Cohen M L 1982 Phys. Rev. B 26 1738
- [35] Gonze X, Stumpf R and Scheffler M 1991 Phys. Rev. B 44 8503
- [36] Monkhorst H J and Pack J D 1976 Phys. Rev. B 13 5188
- [37] Wei S H and Zhang S B 2001 Phys. Rev. B 63 045112
- [38] Catti M, Valerio G, Dovesi R and Causà M 1994 Phys. Rev. B 49 14 179
- [39] Fischer P 1967 W. Kristallogr. 124 275
- [40] MacDonald J R 1969 Rev. Mod. Phys. 38 669 MacDonald J R 1969 Rev. Mod. Phys. 41 316
- [41] Anderson O L 1995 Equations of State of Solids for Geophysics and Ceramic Science (New York: Oxford University Press)
- [42] Brennan B J and Stacey F D 1979 J. Geophys. Res. 84 5535
- [43] Vinet P, Ferrante J, Smith J R and Rose J H 1987 Phys. Rev. B 35 1945
- [44] Shanker J, Kushwah S S and Kumar P 1997 Physica B 239 337
- [45] Birch F 1986 J. Geophys. Res. 91 4949
- [46] Murnaghan F D 1944 Proc. Natl Acad. Sci. USA 30 244
- [47] Mo S D and Ching W Y 1996 Phys. Rev. B 54 16 555
- [48] Chang Z P and Barsch G R 1973 J. Geophys. Res. 78 2418
- [49] Yoneda A 1990 J. Phys. Earth 38 19
- [50] Kruger M B, Nguyen J H, Caldwell W and Jeanloz R 1997 Phys. Rev. B 56 1
- [51] Fuchs M, Bockstedte M, Pehlke E and Scheffler M 1998 Phys. Rev. B 57 2134
- [52] Lauwers H A and Herman M A 1980 J. Phys. Chem. Solids 41 223
- [53] Gonze X and Lee C 1997 Phys. Rev. B 55 10355
- [54] Palik E D (ed) 1991 Handbook of Optical Constants of Solids vol 2 (London: Academic)
- [55] Shannon R D and Rossman G R 1991 J. Phys. Chem. Solids 52 1055
- [56] Thibaudeau P and Da Rocha S, in preparation
- [57] Baroni S and Resta R 1986 Phys. Rev. B 33 7017
- [58] Cynn H, Anderson O L and Nicol N 1993 Pure Appl. Geophys. (PAGEOPH) 141 415
- [59] McMillan P F and Hofmeister A M 1988 Rev. Mineral. 18 99
- [60] Barker A S 1963 Phys. Rev. 132 1474
- [61] Kurosawa T J 1961 J. Phys. Soc. Japan 16 1298
- [62] Scott J F 1971 Phys. Rev. B 4 1360
- [63] Gervais F 1976 Solid State Commun. 18 191
- [64] Gervais F and Arend H 1983 Z. Phys. B 50 17
- [65] Gervais F and Kaczmarek W 1983 Z. Phys. B 51 137
- [66] Tropf W J, Thomas M E and Haris T J 1995 Properties of crystals and glasses Handbook of Optics, Devises, Measurements and Properties 2nd edn, vol 2, ed M Bass, E W Van Stryland, D R Williams and W L Wolfe (New York: McGraw-Hill) ch 33
- [67] Striefler M E and Boldish S I 1978 J. Phys. C: Solid State Phys. 11 L237
- [68] Fraas L M, Moore J E and Salzberg J B 1973 J. Chem. Phys. 58 3585
- [69] O'Horo M P, Frisillo A L and White W B 1973 J. Phys. Chem. Solids 34 23
- [70] Preudhomme J and Tarte P 1971 Spectrochim. Acta A 27 1817